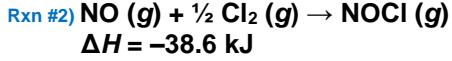
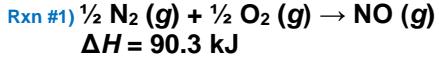


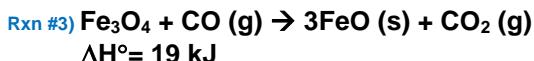
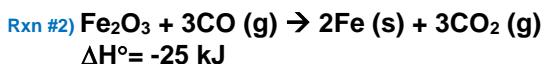
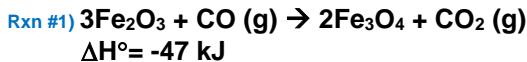
### Hess's Law #1

#	Reaction	$\Delta H^\circ$
1	$C + 2H_2 \rightarrow CH_4$	-74.80 kJ
2	$C + O_2 \rightarrow CO_2$	-393.50 kJ
3	$H_2 + \frac{1}{2}O_2 \rightarrow H_2O$	-285.83 kJ

### Hess's Law #2



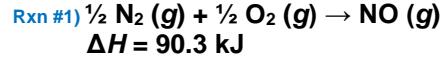
### Hess's Law #3



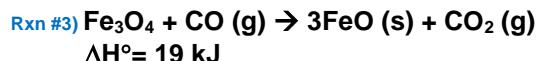
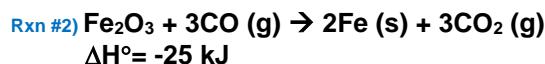
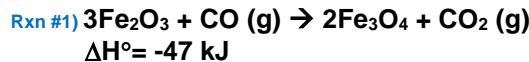
### Hess's Law #1

#	Reaction	$\Delta H^\circ$
1	$C + 2H_2 \rightarrow CH_4$	-74.80 kJ
2	$C + O_2 \rightarrow CO_2$	-393.50 kJ
3	$H_2 + \frac{1}{2}O_2 \rightarrow H_2O$	-285.83 kJ

### Hess's Law #2



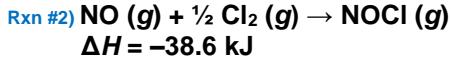
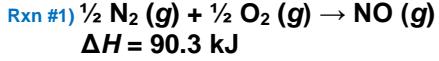
### Hess's Law #3



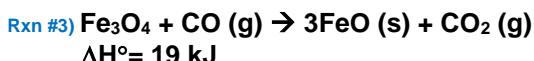
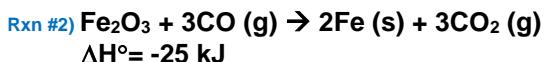
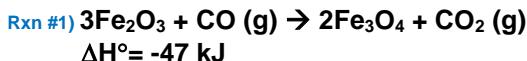
### Hess's Law #1

#	Reaction	$\Delta H^\circ$
1	$C + 2H_2 \rightarrow CH_4$	-74.80 kJ
2	$C + O_2 \rightarrow CO_2$	-393.50 kJ
3	$H_2 + \frac{1}{2}O_2 \rightarrow H_2O$	-285.83 kJ

### Hess's Law #2



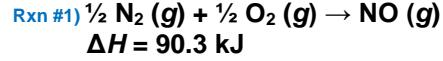
### Hess's Law #3



### Hess's Law #1

#	Reaction	$\Delta H^\circ$
1	$C + 2H_2 \rightarrow CH_4$	-74.80 kJ
2	$C + O_2 \rightarrow CO_2$	-393.50 kJ
3	$H_2 + \frac{1}{2}O_2 \rightarrow H_2O$	-285.83 kJ

### Hess's Law #2



### Hess's Law #3

